

# Vierodt/Simultaneous Equation Method

## Procedure:

### Step 1: Preparation of standard solution of Telmisartan

- i. Standard stock solution of Telmisartan is prepared by dissolving 10 mg of Telmisartan drug in a 10 mL volumetric flask using methanol (If drug is not dissolved, sonicate it for 15 minutes). Stock solutions of 1000  $\mu\text{g/mL}$  were obtained in this manner.
- ii. Then, take pipette out 1 ml of solution from above prepared 1000  $\mu\text{g}$  solution and add it into 10 ml volumetric flask. Make up the volume up to the mark with pH 7.5 phosphate buffer as a solvent. A solution of 100  $\mu\text{g}$  is prepared.
- iii. From these 100  $\mu\text{g}$  stock solutions, working standard solutions of concentration 5, 10, 15, 20, 25  $\mu\text{g/mL}$  each were prepared by appropriate dilutions.
- iv. Working standard solutions of concentration 10  $\mu\text{g/mL}$  were scanned in the entire UV range to determine the  $\lambda_{\text{max}}$ . Use solution prepared by dissolving (1 ml methanol + 9 ml pH 7.5 phosphate buffer) as a blank solution.
- v. In our case the  $\lambda_{\text{max}}$  of Telmisartan were found to be 295 nm.

### Step 2: Preparation of standard solution of Hydrochlorothiazide

- i. Standard stock solution of Hydrochlorothiazide is prepared by dissolving 10 mg of Hydrochlorothiazide drug in a 10 mL volumetric flask using methanol (If drug is not dissolved, sonicate it for 15 minutes). Stock solutions of 1000  $\mu\text{g/mL}$  were obtained in this manner.
- ii. Then, take pipette out 1 ml of solution from above prepared 1000  $\mu\text{g}$  solution and add it into 10 ml volumetric flask. Make up the volume up to the mark with 0.1 M HCl solution as a solvent. A solution of 100  $\mu\text{g}$  is prepared.
- iii. From these 100  $\mu\text{g}$  stock solutions, working standard solutions of concentration 5, 10, 15, 20, 25  $\mu\text{g/mL}$  each were prepared by appropriate dilutions.
- iv. Working standard solutions of concentration 10  $\mu\text{g/mL}$  were scanned in the entire UV range to determine the  $\lambda_{\text{max}}$ . Use solution prepared by dissolving (1 ml methanol + 9 ml 0.1 M HCl) as a blank solution.
- v. In our case the  $\lambda_{\text{max}}$  of Hydrochlorothiazide were found to be 224 nm.

**Step 3: Take absorbance of all stock solution of Telmisartan**

- i. Measure the absorbance of all the five-stock solution of concentration 5, 10, 15, 20, 25  $\mu\text{g/mL}$  of Telmisartan on UV at both wavelength of 295 nm and 224 nm. Use solution prepared by dissolving (1 ml methanol + 9 ml pH 7.5 phosphate buffer) as a blank solution.
- ii. Note the absorbance of all the stock solutions of Telmisartan in below table.

| <b>Table 1: Absorbance of stock solutions of Telmisartan</b> |  |  |
|--|--|--|
| <b>Conc.</b>   | <b>Absorbance at <math>\lambda_1 = 295 \text{ nm}</math></b> | <b>Absorbance at <math>\lambda_2 = 224 \text{ nm}</math></b> |
| 5 $\mu\text{g/mL}$   | 0.221  | 0.416  |
| 10 $\mu\text{g/mL}$  | 0.332  | 0.731  |
| 15 $\mu\text{g/mL}$  | 0.635  | 1.152  |
| 20 $\mu\text{g/mL}$  | 0.831  | 1.567  |
| 25 $\mu\text{g/mL}$  | 1.013  | 1.895  |

**Step 4: Take absorbance of all stock solution of Hydrochlorothiazide**

- i. Measure the absorbance of all the five-stock solution of concentration 5, 10, 15, 20, 25  $\mu\text{g/mL}$  of Hydrochlorothiazide on UV at both wavelength of 295 nm and 224 nm. Use solution prepared by dissolving (1 ml methanol + 9 ml 0.1 M HCl) as a blank solution.
- ii. Note the absorbance of all the stock solutions of Hydrochlorothiazide in below table.

| <b>Table 2: Absorbance of stock solutions of Hydrochlorothiazide</b> |  |  |
|--|--|--|
| <b>Conc.</b>   | <b>Absorbance at <math>\lambda_1 = 295 \text{ nm}</math></b> | <b>Absorbance at <math>\lambda_2 = 224 \text{ nm}</math></b> |
| 5 $\mu\text{g/mL}$   | 0.055  | 0.453  |
| 10 $\mu\text{g/mL}$  | 0.06   | 0.55   |
| 15 $\mu\text{g/mL}$  | 0.119  | 0.845  |
| 20 $\mu\text{g/mL}$  | 0.177  | 1.131  |
| 25 $\mu\text{g/mL}$  | 0.226  | 1.539  |

**Step 5: Calculate the absorptivity**

- i. Convert the concentration of stock solution from  $\mu\text{g/mL}$  to  $\mu\text{g}/100 \text{ mL}$ . Then convert it from  $\mu\text{g}/100 \text{ mL}$  to  $\text{g}/100 \text{ mL}$ .

For example,

$$10 \mu\text{g/mL} \longrightarrow 1000 \mu\text{g}/100 \text{ mL} \longrightarrow 0.001 \text{ g}/100 \text{ mL}$$

- ii. Then calculate the absorptivity from the absorbance by using the below formula. (Use molar concentration in gram to calculate the absorptivity)

$$A = \epsilon cl$$

|            |                              |                               |
|------------|------------------------------|-------------------------------|
| $A$        | Absorbance                   |                               |
| $\epsilon$ | Molar absorption coefficient | $\text{M}^{-1}\text{cm}^{-1}$ |
| $C$        | Molar concentration          | M                             |
| $l$        | optical path length          | cm                            |

For example,

$$\begin{aligned} \text{Absorptivity} &= \text{Absorbance}/\text{Concentration} \\ &= 0.332/0.001 \\ &= 332 \end{aligned}$$

| Table 3: Absorptivity (Telmisartan) |  |  |
|-------------------------------------|--|--|
| Conc.                               | Absorptivity at $\lambda_1 = 295 \text{ nm}$ | Absorptivity at $\lambda_2 = 224 \text{ nm}$ |
| 5 $\mu\text{g/mL}$                  | 44.2   | 83.2   |
| 10 $\mu\text{g/mL}$                 | 332  | 731  |
| 15 $\mu\text{g/mL}$                 | 423.3333333                                  | 768  |
| 20 $\mu\text{g/mL}$                 | 415.5  | 783.5  |
| 25 $\mu\text{g/mL}$                 | 405.2  | 758  |

| Table 4: Absorptivity (Hydrochlorothiazide) |  |  |
|---|--|--|
| Conc.                                       | Absorptivity at $\lambda_1 = 295 \text{ nm}$ | Absorptivity at $\lambda_2 = 224 \text{ nm}$ |
| 5 $\mu\text{g/mL}$                          | 11   | 90.6   |
| 10 $\mu\text{g/mL}$                         | 60   | 550  |
| 15 $\mu\text{g/mL}$                         | 79.33333333                                  | 563.3333333                                  |
| 20 $\mu\text{g/mL}$                         | 88.5   | 565.5  |
| 25 $\mu\text{g/mL}$                         | 90.4   | 615.6  |

**Step 6: Take the absorbance of marketed tablet**

- i. Purchase a tablet from market that contain the same amount of drug as of your formulated tablet. (For example, your formulated tablet contains 40 mg Telmisartan and 12.5 mg Hydrochlorothiazide then purchase a marketed tablet that should also contain 40 mg Telmisartan and 12.5 mg Hydrochlorothiazide)
- ii. The marketed tablet is then crushed to a fine powder. An accurately weighed powder sample equivalent to 25 mg of Telmisartan was transferred to a 25 ml volumetric flask.
- iii. Calculate the equivalent weight by using below formula.

$$\text{Equivalent Weight} = \frac{\text{Quantity Required} \times \text{Average Weight of Tablet}}{\text{Label Claim}}$$

- iv. Then volume was made up to the mark with methanol solution and sonicate for 15 minutes to ensure the powder dissolves properly. The concentration of prepared solution will be of 1000 µg/mL.
- v. The solution was then filtered through Whatman filter paper no. 41.
- vi. Then pipette out 1 ml from filtered solution and make volume with pH 7.5 phosphate buffer solution to produce a stock solution of concentration 100 µg/mL.
- vii. Again, pipette out 2 ml from above solution and volume was make up with pH 7.5 phosphate buffer solution to prepare a stock solution of concentration 20 µg/mL.
- viii. Measure the absorbance of stock solution of concentration 20 µg/mL on UV at both wavelength of 295 nm and 224 nm. Use solution prepared by dissolving (1 ml methanol + 9 ml pH 7.5 phosphate buffer) as a blank solution.
- ix. Note the readings in below table.

| <b>Table 5: Absorbance of stock solutions of marketed tablet</b> |   |   |
|--|---|---|
| <b>Conc.</b>   | <b>Absorbance at λ<sub>1</sub> = 295 nm</b> | <b>Absorbance at λ<sub>2</sub> = 224 nm</b> |
| 20 µg/mL   | 0.895                                       | 1.871                                       |

**Step 7: Determine the ax1, ax2, ay1 and ay2**

- i. Determine the ax1, ax2, ay1 and ay2 from the table 3 and table 4.
- ii. Were,  
ax1 = Absorptivity of Telmisartan at  $\lambda_1$   
ax2 = Absorptivity of Telmisartan at  $\lambda_2$   
ay1 = Absorptivity of Hydrochlorothiazide at  $\lambda_1$   
ay2 = Absorptivity of Hydrochlorothiazide at  $\lambda_2$

**Step 8: Determine the A<sub>1</sub> and A<sub>2</sub>**

- i. Determine the A<sub>1</sub> and A<sub>2</sub> from the table 5.
- ii. Were,  
A<sub>1</sub> = Absorbance at  $\lambda_1$   
A<sub>2</sub> = Absorbance at  $\lambda_2$

**Step 9: Determine the Cx and Cy**

- i. Determine the Cx = Concentration of Telmisartan and Cy = Concentration of Hydrochlorothiazide by using the below formula.

$$Cx = \frac{A_2 a_{y1} - A_1 a_{y2}}{a_{x2} a_{y1} - a_{x1} a_{y2}}$$
$$Cy = \frac{A_1 a_{x2} - A_2 a_{x1}}{a_{x2} a_{y1} - a_{x1} a_{y2}}$$

- ii. The founded value of Cx will be in g/100 ml. So, convert the value in g/10 ml and then to mg/10 ml.

For example,

The founded value of Cx = 0.001986518 g/100 ml  
0.000198652 g/10 ml  
0.198651755 mg/10 ml

- iii. It means 0.198651755 mg of Telmisartan drug is present in 2 ml of stock solution. Now find how much amount of drug should present in 10 ml by doing cross multiplication.

|     |                            |   |           |
|-----|----------------------------|---|-----------|
|     | 2                          | → | 0.1986518 |
|     | 10                         | → | x         |
| X = | $\frac{0.1090097 * 10}{2}$ |   |           |
| X = | 0.993259                   |   |           |

- iv. Multiply the founded X value with 25 to obtain the drug present in 25 ml of stock solution.

|     |                          |    |
|-----|--------------------------|----|
| X = | 0.993259                 |    |
|     | x                        | 25 |
|     | <b>FINAL = 24.831475</b> |    |

- v. Follow the step ii, iii and iv for obtaining the concentration of hydrochlorothiazide drug.